

GRADE 8
Science

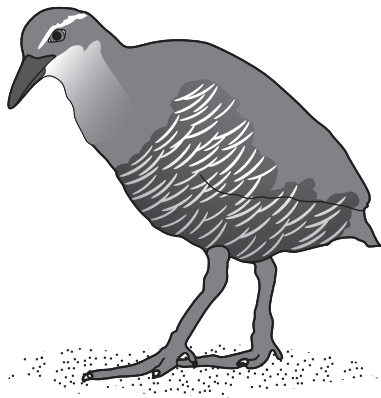
Administered April 2015

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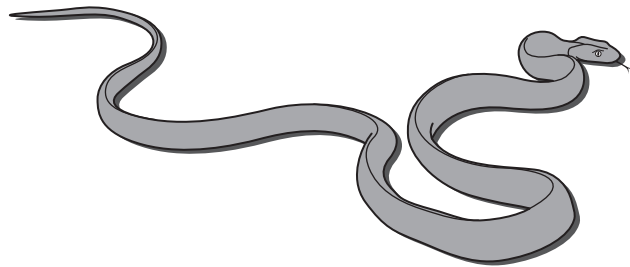
SCIENCE

Read each question carefully. For a multiple-choice question, determine the best answer to the question from the four answer choices provided. For a griddable question, determine the best answer to the question. Then fill in the answer on your answer document.

- 1** The Guam rail is a flightless bird found on the island of Guam. It builds a shallow nest on the ground and lays eggs throughout the year. There are no native species of snakes on Guam, so the Guam rail had no experience with snake predators until the brown tree snake was accidentally introduced sometime between 1944 and 1952. These snakes, which eat a wide variety of animal species, quickly established a large population on Guam.



Guam rail

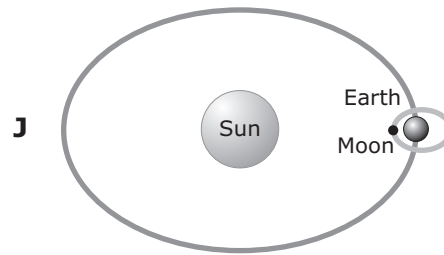
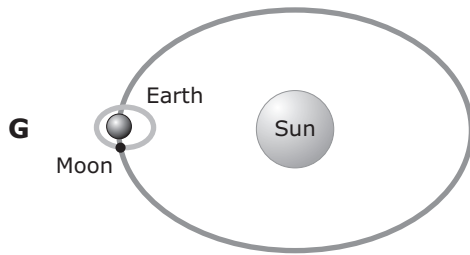
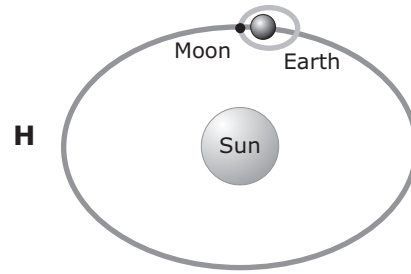
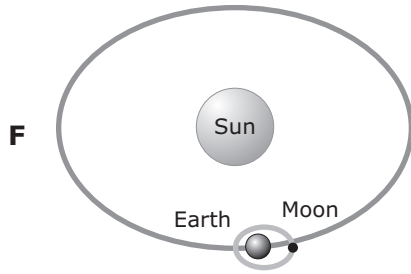


Brown tree snake

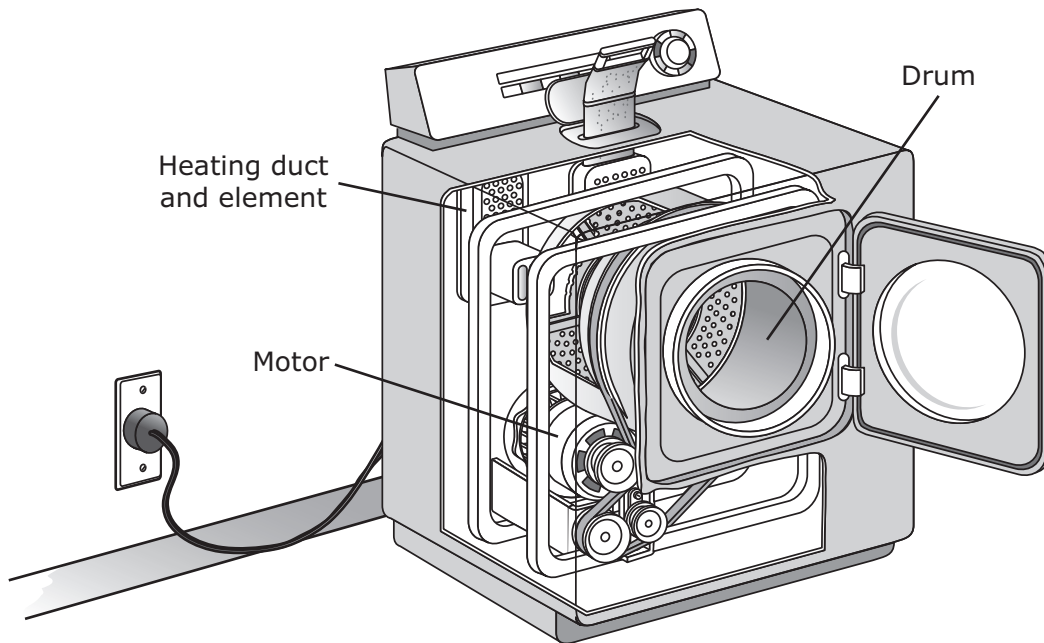
How was the Guam rail most likely affected by the introduction of the brown tree snake?

- A** The brown tree snake ate only species that lived in trees, so it had no effect on the Guam rail.
- B** The Guam rail quickly developed the ability to fly in order to escape from the brown tree snake.
- C** The population of the Guam rail declined, and the species nearly became extinct.
- D** The Guam rail began to build its nests in trees.

- 2 Different relative positions of the sun, the moon, and Earth have different effects on Earth's oceans. Which model shows the positions of the sun, the moon, and Earth that have the greatest effect on ocean tides?



3 The main parts of a working clothes dryer are shown in the diagram.



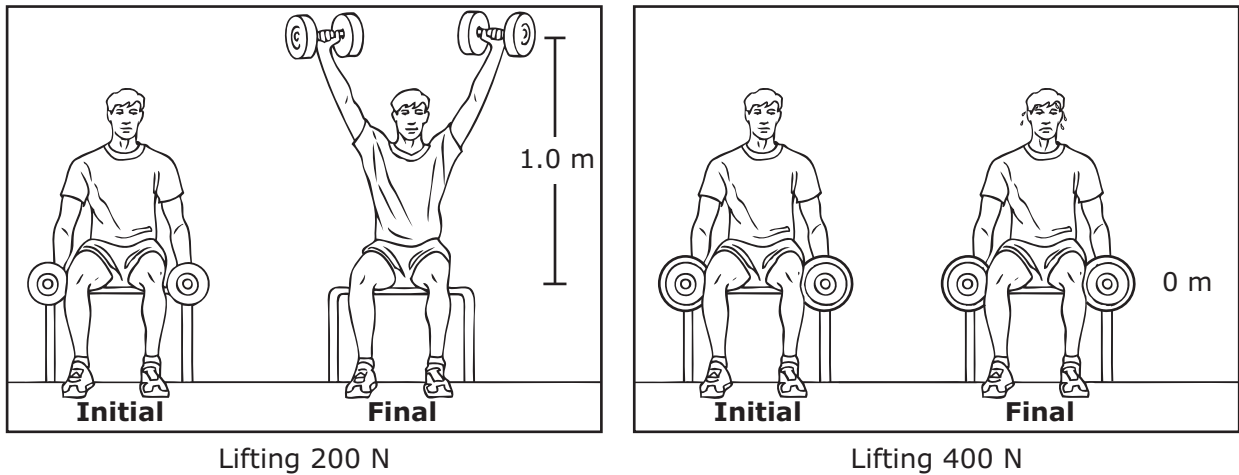
This appliance dries clothes primarily by converting —

- A electrical energy to heat and mechanical energy
- B mechanical energy to heat and electrical energy
- C electrical energy to heat and light
- D mechanical energy to heat and vaporization

4 Which list of characteristics describes organisms classified as animals?

- F Unicellular, prokaryotic, autotrophic
- G Multicellular, eukaryotic, heterotrophic
- H Unicellular, eukaryotic, heterotrophic
- J Multicellular, eukaryotic, autotrophic

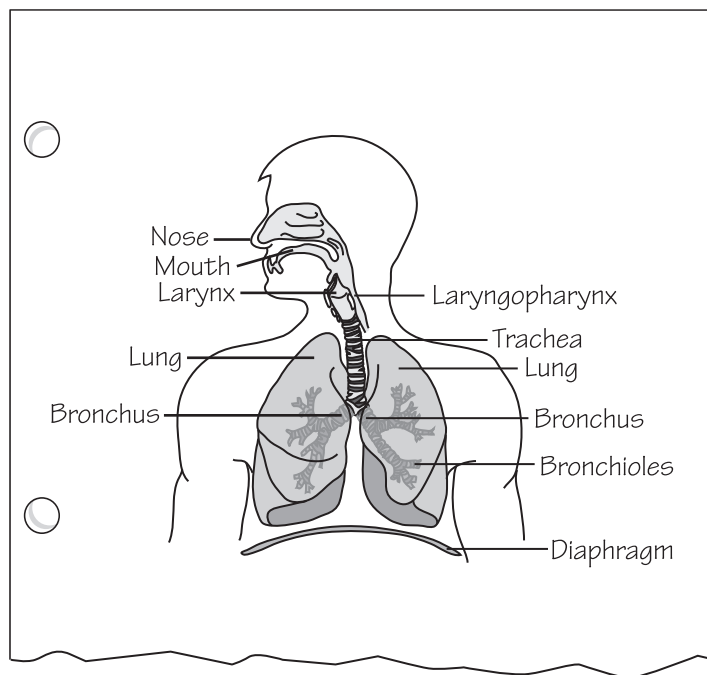
- 5 An athlete is trying to lift some weights. He successfully lifts 200 N weights, but when he tries to lift 400 N weights, he is unsuccessful.



Which of these correctly compares the amount of work the athlete does on the weights?

- A The athlete does no work on either the 200 N weights or the 400 N weights.
- B The athlete does no work on the 200 N weights but does work on the 400 N weights.
- C The athlete does work on both the 200 N weights and the 400 N weights.
- D The athlete does work on the 200 N weights but does no work on the 400 N weights.

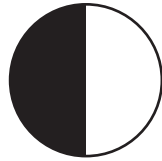
- 6 A student drew the picture of a human body system shown below. The student labeled some of the organs of the system.



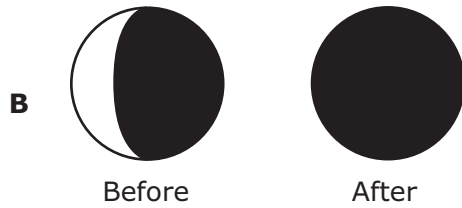
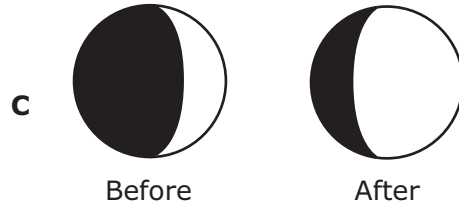
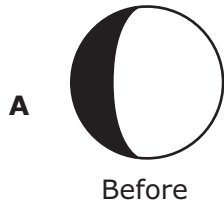
The main function of this human body system is to —

- F sense odors in the environment
- G exchange gases with the environment
- H chew food as it enters the mouth
- J trap particles that enter the nose

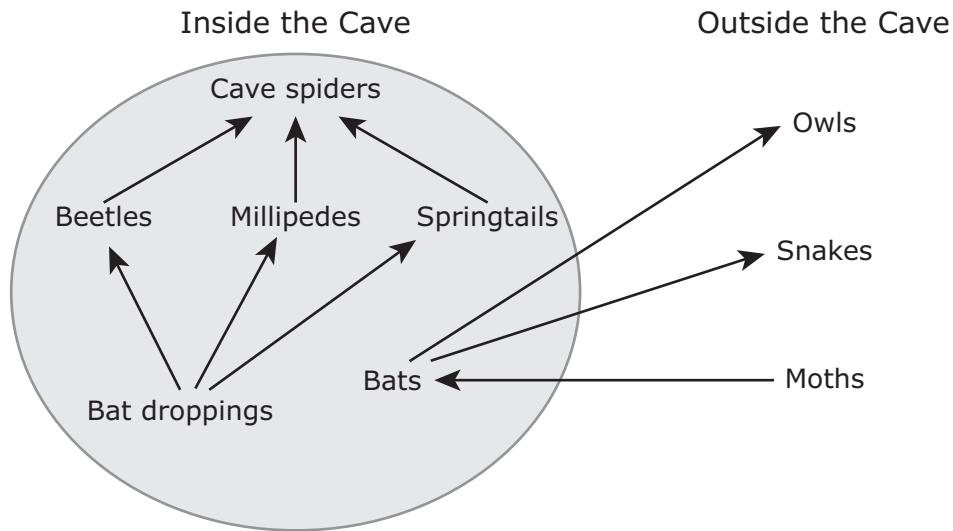
7 A phase of the moon is shown below.



Which pair of diagrams shows the moon's phase immediately before and immediately after this phase?



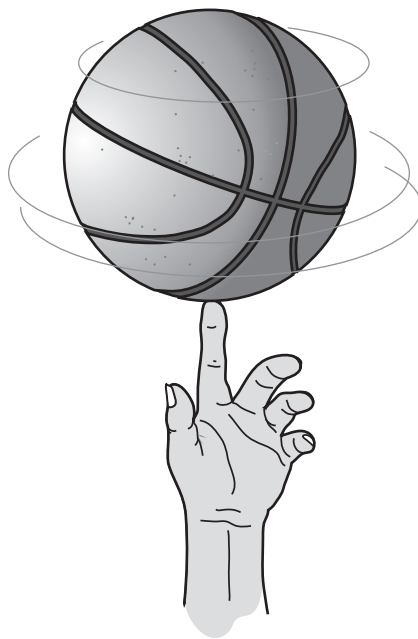
- 8 Many Texas caves provide a habitat for animals that live in total darkness. The caves are damp and cool, and most of the caves are rich in the mineral calcite. The diagram below shows the relationships of some organisms involved in a cave ecosystem.



Based on the diagram, which of the following statements is valid?

- F** Owls are predators of bats, and bats are predators of moths.
- G** Beetles compete with bats and cave spiders.
- H** Snakes prey on bats and owls.
- J** None of the above
-
- 9 Coal contains carbon and other elements. Carbon dioxide forms when coal burns in the presence of oxygen. Which of these is the best evidence that a chemical reaction occurs when coal burns?
- A** The shape of the coal changes.
- B** Oxygen is present.
- C** A new substance is produced.
- D** Coal is made up of more than one element.

- 10** A student wants to use a spinning basketball to model the day-and-night cycle on Earth.



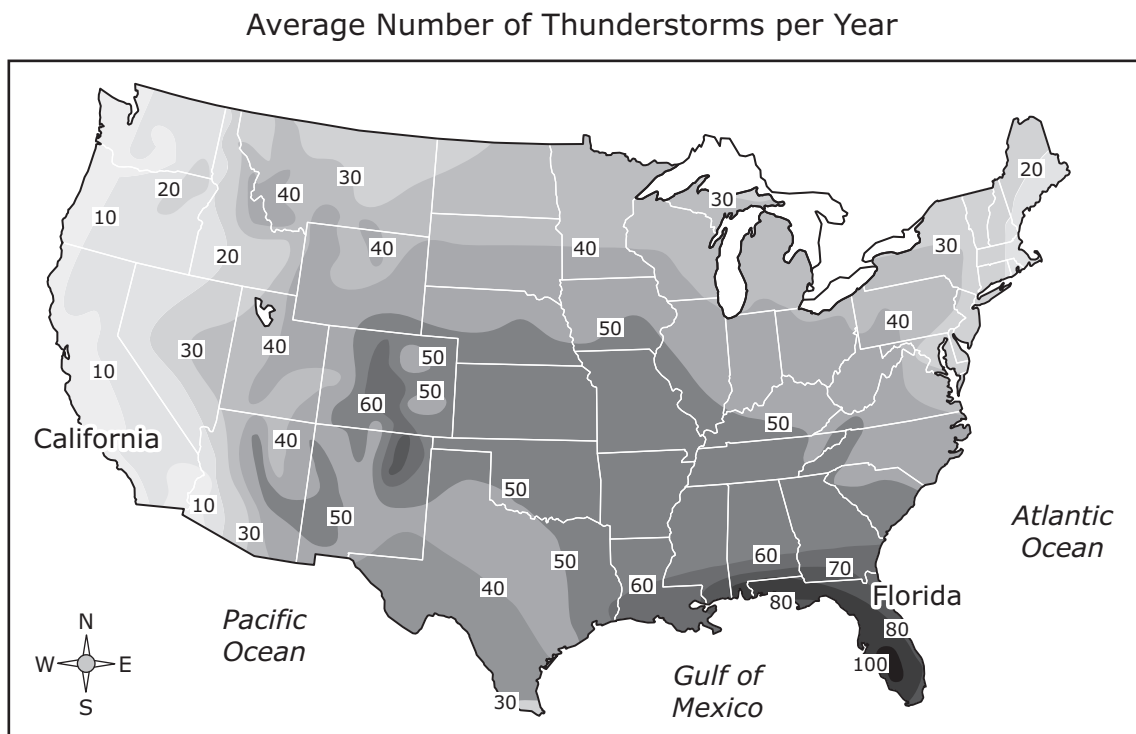
What should the student do in order to best model the cause of the day-and-night cycle?

- F** Spin the ball in different directions
 - G** Tilt the ball while it is spinning
 - H** Paint one half of the ball black and the other half white
 - J** Shine a bright light on one side of the ball
-
- 11** The atoms of a certain element each contain 11 protons and 1 valence electron. Which statement correctly identifies this element and describes its chemical reactivity?
- A** The element is sodium, and it is highly reactive.
 - B** The element is fluorine, and it is not very reactive.
 - C** The element is sodium, and it is not very reactive.
 - D** The element is fluorine, and it is highly reactive.

12 A bus travels 20 km in 30 minutes. What is the average speed of the bus?

- F** 20 km/h
- G** 30 km/h
- H** 40 km/h
- J** 50 km/h

13 Thunderstorms are more common in some areas than in others. The map shows the average number of thunderstorms per year in different parts of the United States.



Source: NOAA National Weather Service

Why are there significantly more thunderstorms in Florida than in California?

- A** The state of Florida is smaller than the state of California.
- B** The air in Florida is more stable than the air in California.
- C** The air above Florida holds less moisture than the air above California.
- D** The ocean currents near Florida are warmer than the ocean currents near California.

- 14** During a demonstration of Newton's laws of motion, a student used the setup shown in Figure 1. The student flicked the index card with a fingertip, and the coin fell straight down into a plastic cup as shown in Figure 2.

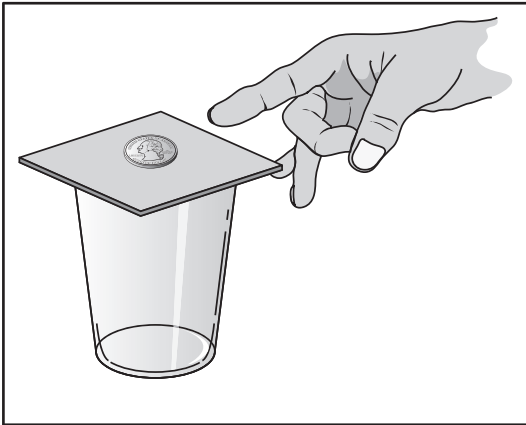


Figure 1

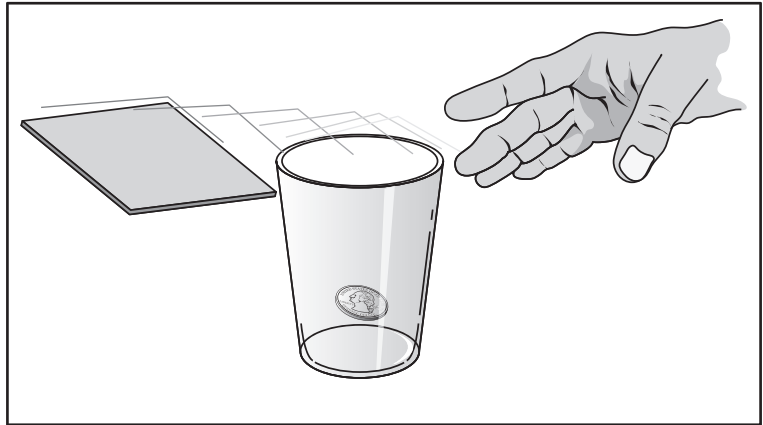


Figure 2

Which of these best explains why the coin fell straight down into the cup instead of remaining on the index card?

- F** The coin was at rest until the card was removed, so it tended to remain in the same location. Once the card was gone, the unbalanced force of gravity caused the coin to fall.
- G** Moving the card applied an action force on the coin. Since the card was gone, gravity applied a reaction force on the coin.
- H** The card had less mass than the coin, so a smaller force of gravity acted on the card. The larger force of gravity on the coin made it fall.
- J** The acceleration of the coin falling into the cup was equal and opposite to the acceleration of the card.

- 15 For an investigation a student records data about four unknown substances.

Data for Unknown Substances

Substance	Mass (g)	Volume (cm ³)	Density (g/cm ³)
1	6.95	4.0	
2	4.54	2.0	
3	5.40	3.0	
4	10.35	5.0	

The student then calculates the densities of the unknown substances and compares them with the table of densities of known substances shown below.

Densities of Some Known Substances

Substance	Density (g/cm ³)
Calcium	1.54
Carbon	2.27
Magnesium	1.74
Phosphorus	1.82
Platinum	21.46
Sulfur	2.07

Which unknown substance is most likely carbon?

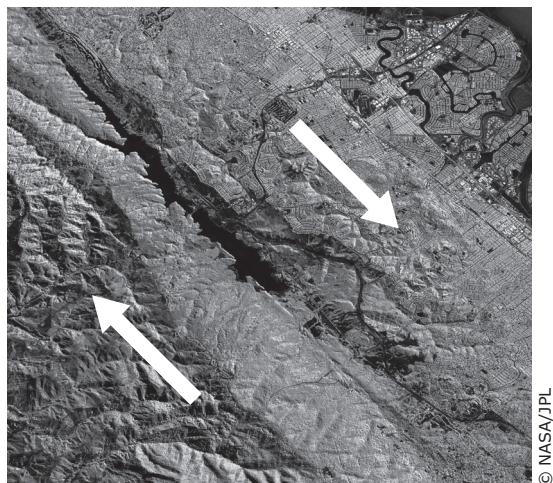
- A Substance 1
- B Substance 2
- C Substance 3
- D Substance 4

- 16** Which of these human activities is most likely to cause the excessive growth of phytoplankton in the world's oceans?
- F** The use of chemical fertilizers that are carried by runoff into rivers
 - G** The logging of old-growth forests, which results in erosion
 - H** The spraying of chemical herbicides that reduce carbon dioxide in the air
 - J** The mining of fossil fuels, which requires digging underground tunnels

-
- 17** An atom of a certain element has 36 protons, 36 electrons, and a mass number of 84. At room temperature this element is a very stable gas. How many neutrons are in this atom?

Record your answer and fill in the bubbles on your answer document. Be sure to use the correct place value.

- 18** The image shows a plate boundary. Arrows have been added to indicate the movement of the plates.



What plate boundary feature is also shown?

- F** A ridge formed by an inactive convergent boundary
- G** A valley forming at an active divergent boundary
- H** A fault forming at an active transform boundary
- J** A volcano formed by an active transform boundary

19 Four students push carts filled with sports equipment across the gym. Each student pushes with the same amount of force. Which cart has the greatest change in speed?



12 tennis rackets
 $m = 10 \text{ kg}$



18 exercise mats
 $m = 20 \text{ kg}$

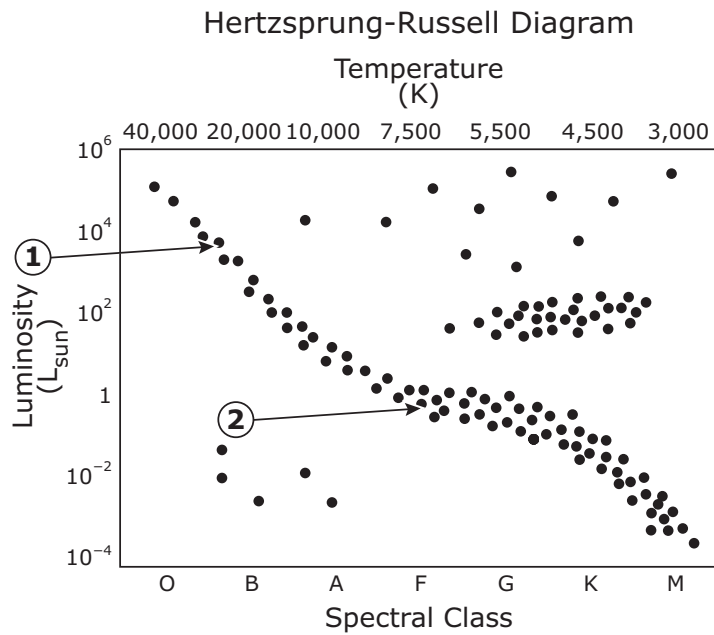


15 sports balls
 $m = 15 \text{ kg}$



28 pieces of rope
 $m = 25 \text{ kg}$

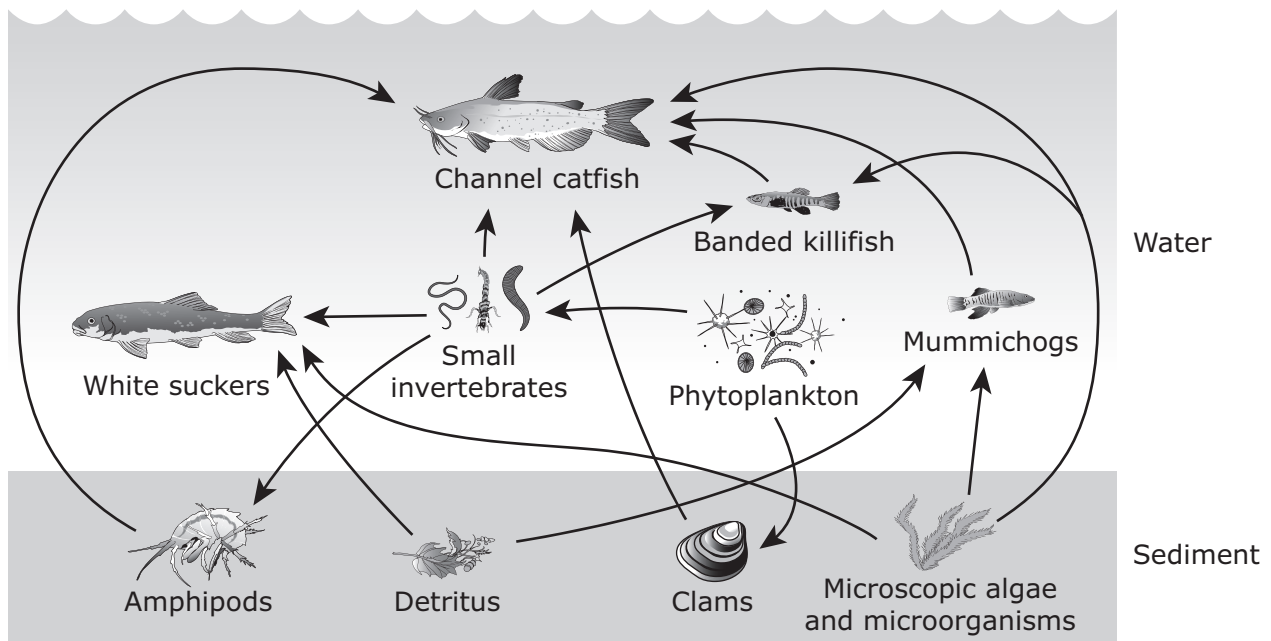
20 Two stars are identified on the Hertzsprung-Russell diagram below.



Based on this diagram, how do the characteristics of Star 1 and Star 2 compare?

- F** Star 1 is cooler and less bright than Star 2.
- G** Star 1 is hotter and brighter than Star 2.
- H** Star 1 is cooler and brighter than Star 2.
- J** Star 1 is hotter and less bright than Star 2.

- 21 The partial food web shown below is found in an aquatic environment. This type of environment has many organisms in the same trophic levels.

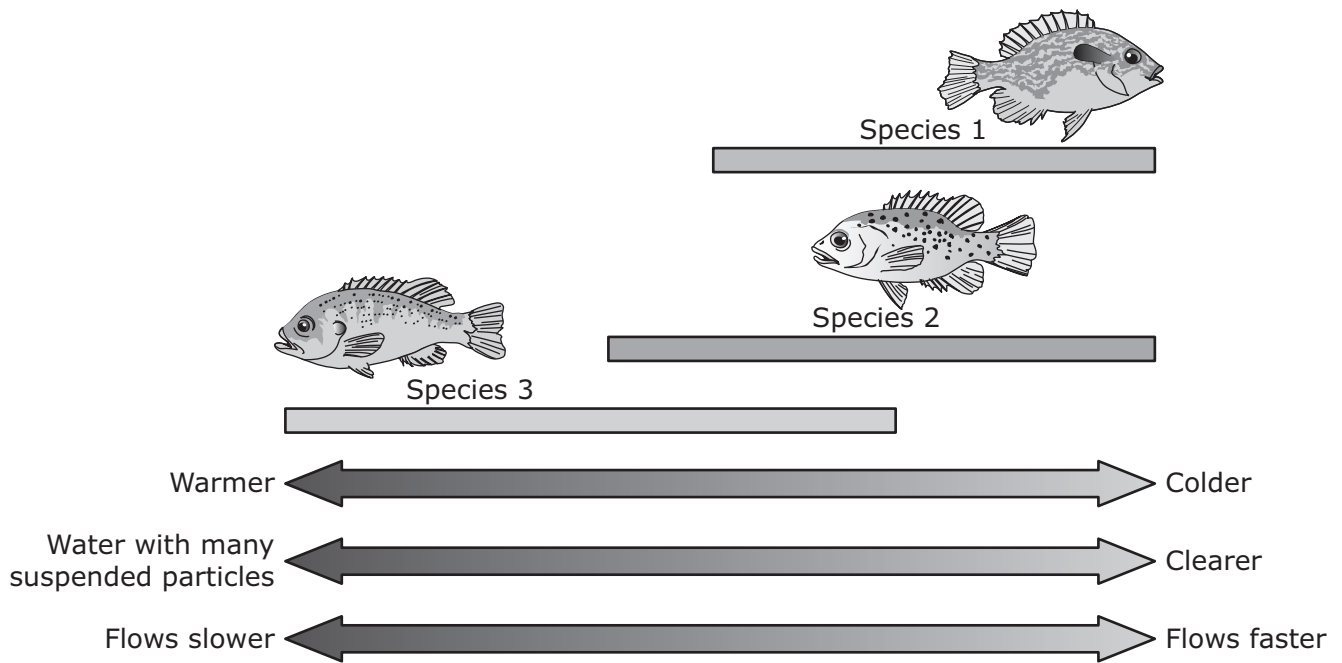


Which two types of organisms in this aquatic food web have a producer–consumer relationship?

- A Microscopic algae and white suckers
 - B Small invertebrates and amphipods
 - C Phytoplankton and mummichogs
 - D Amphipods and channel catfish
-
- 22 All the following reactions are correctly balanced **except** —

- F $2\text{C}_2\text{H}_2 + 5\text{O}_2 \longrightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$
- G $4\text{Fe} + 3\text{O}_2 \longrightarrow 2\text{Fe}_2\text{O}_3$
- H $\text{NaHCO}_3 \longrightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$
- J $2\text{CH}_3\text{OH} + 3\text{O}_2 \longrightarrow 2\text{CO}_2 + 4\text{H}_2\text{O}$

23 The graph below shows the range of water conditions preferred by three species of fish in an Ozark stream.



Water with which of these conditions would cause the most competition for resources among all three species of fish?

- A** Very cold, very clear, and very fast
- B** Very warm, very clear, and very slow
- C** Mild temperatures, some suspended particles, flows very fast
- D** Mild temperatures, some suspended particles, flows at a medium speed

- 24** The periodic table is organized into groups and periods of elements. The characteristics of a certain group of elements are listed below.

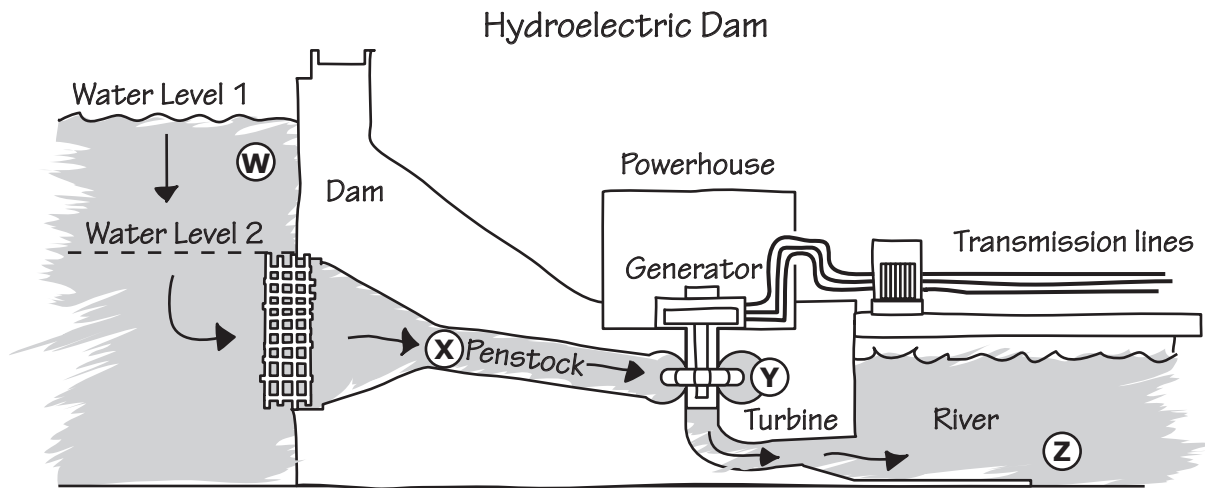
Characteristics of a Group of Elements

- Is shiny
- Is solid at room temperature
- Has atoms with two valence electrons

Which of these elements is in this group?

- F** Lithium
- G** Strontium
- H** Aluminum
- J** Silicon

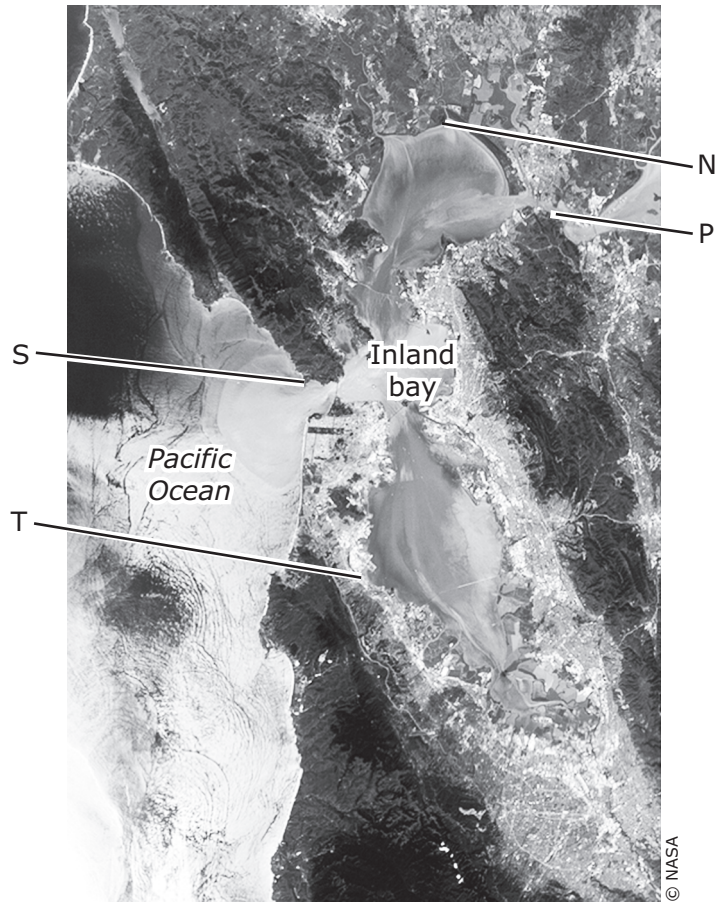
- 25 A student drew the diagram below to show the movement of water through a hydroelectric dam.



The student used the diagram to describe changes in the potential and kinetic energy of the water. At which location is the gravitational potential energy of the water the greatest?

- A Location W
- B Location X
- C Location Y
- D Location Z

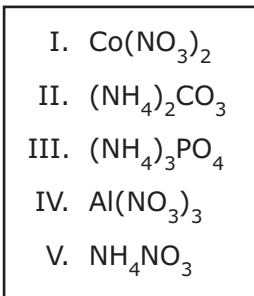
- 26** The satellite photo below shows San Francisco, California, which has many miles of coastline and an inland bay.



Which area of the coastline has most likely experienced the greatest effect of erosion from waves over hundreds of years?

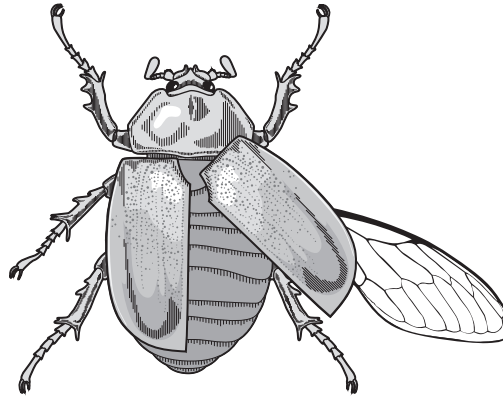
- F** Area N
- G** Area P
- H** Area S
- J** Area T

27 Which of these formulas contain equal numbers of nitrogen atoms?



- A Formulas I and III
- B Formulas I and IV
- C Formulas II and III
- D Formulas I, II, and V

28 A student examines the winged insect shown below.



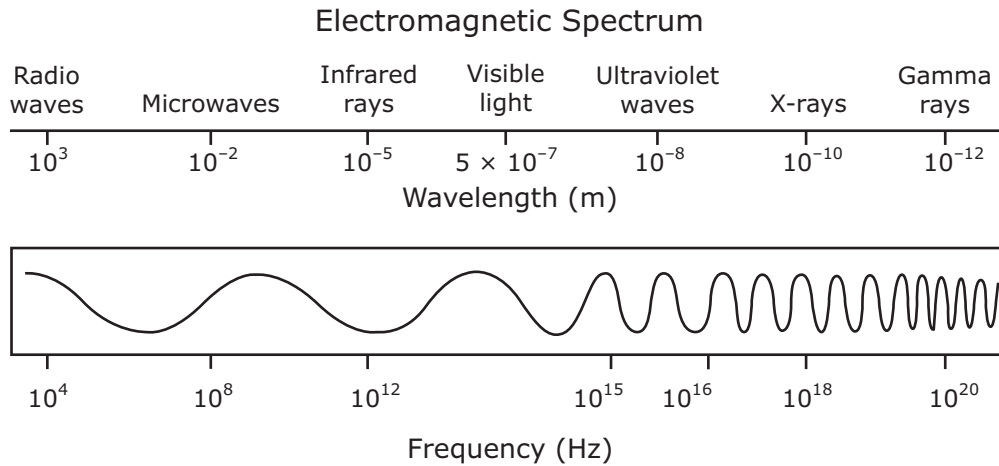
Dichotomous Key

Step	Characteristics	Order
1a	Insect has an extremely long prothorax (neck)	Go to 2
1b	Insect has a short or no prothorax	Go to 3
2a	Forelegs come together in a "praying" position	Mantodea
2b	Forelegs do not come together in a "praying" position	Raphidioptera
3a	Wings are armor-like with membranous hind wings underneath	Coleoptera
3b	Wings are not armor-like	Go to 4
4a	Wings are triangular in shape	Go to 5
4b	Wings are not triangular in shape and head is elongated	Mecoptera
5a	Insect lacks a proboscis (long, slender snout) and has long filaments at abdominal tip	Ephemeroptera
5b	Insect has a proboscis and lacks long filaments at abdominal tip	Lepidoptera

Based on the dichotomous key, in what order should this insect be classified?

- F Mantodea
- G Raphidioptera
- H Coleoptera
- J Lepidoptera

- 29** Quasars are extremely distant celestial bodies. Investigators using a special telescope determined that a certain quasar was emitting waves with a frequency of 1.41×10^9 Hz. An electromagnetic spectrum is shown.



The investigators were most likely using a telescope that detects —

- A** microwaves
- B** visible light
- C** x-rays
- D** gamma rays

- 30** A basketball with a mass of 0.60 kg is accelerated with a force of 10.8 N. If resisting forces are ignored, what is the acceleration of the basketball to the nearest m/s^2 ?

Record your answer and fill in the bubbles on your answer document. Be sure to use the correct place value.

- 31** Scientists are studying the effects of climate change on the Baltic Sea. Some scientists predict that if current trends continue, the Baltic Sea will become less salty, and it will contain less dissolved oxygen. As a result of these changes, the Baltic Sea will have less biodiversity. The location of the Baltic Sea is shown on the map below.



Which statement best explains why biodiversity in the Baltic Sea could decrease if these changes occur?

- A** Fewer types of plants and animals are found in warm water than in cool water.
- B** Plants and animals adapted for living in saltwater will grow more slowly in water that has less salt.
- C** Plants and animals adapted for living in saltier conditions will not be able to survive and reproduce.
- D** Animals adapted for living in warm water will eat all the plants and animals adapted for living in cool water.

- 32** On July 4, 1776, the American colonies declared their independence. The table shows the lunar phases for June 1776.

Lunar Phases for
June 1776

Date	Lunar phase
June 2	Full moon
June 9	Last-quarter moon
June 16	New moon
June 24	First-quarter moon

Source: NASA

What lunar phase occurred on July 4, 1776?

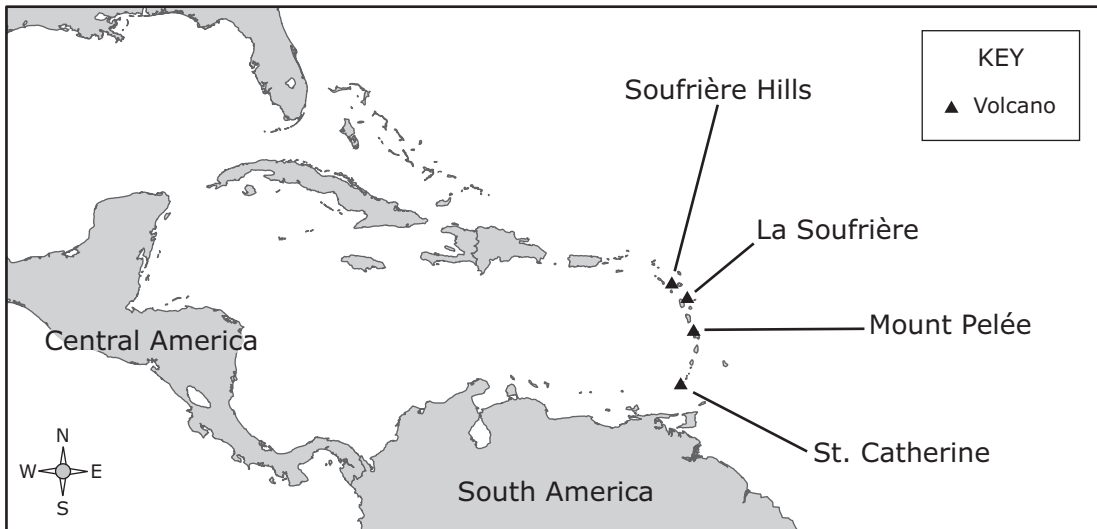
- F** Waning gibbous
- G** Waxing gibbous
- H** Waning crescent
- J** Waxing crescent

-
- 33** Aluminum, sulfur, and chlorine are elements found on the periodic table. Tests involving these three elements would show that at normal room temperature —

- A** sulfur is malleable and aluminum conducts electricity
- B** chlorine is malleable and conducts electricity
- C** aluminum is malleable and conducts electricity
- D** all three elements are malleable and conduct electricity

34 The map below shows the locations of four active volcanoes in the Caribbean Sea.

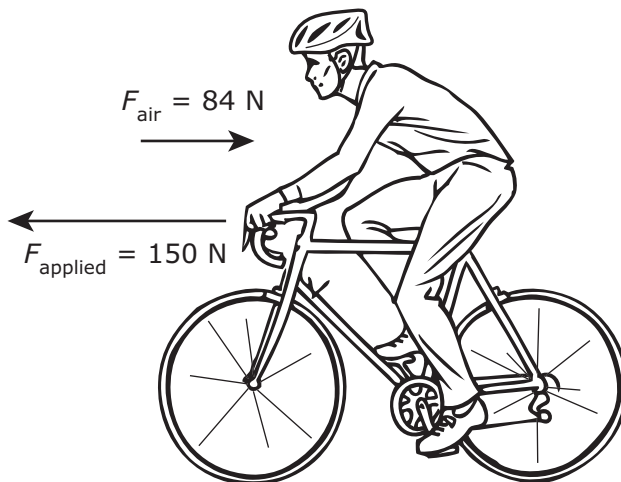
Volcanoes in the Caribbean



What do these volcanoes indicate about plate tectonics in the region?

- F** The volcanoes form an island arc, indicating a converging plate boundary.
- G** The volcanoes form a line, indicating a diverging plate boundary.
- H** Two plates are moving past each other, causing faulting and magma formation.
- J** Two plates are moving apart, creating new seafloor.

35 The diagram below shows two different forces acting on a cyclist riding a bicycle.



The total mass of the cyclist and the bicycle is 100.0 kg. Based on this information, what is the acceleration of the cyclist?

- A** 0.66 m/s^2 backward, because the force of the air slows the cyclist down
- B** 0.66 m/s^2 forward, because the applied force is greater than the force of the air
- C** 2.3 m/s^2 backward, because the forces are opposite and not equal
- D** 2.3 m/s^2 forward, because the cyclist's inertia is greater than the force of the air

36 The equation below represents a chemical reaction that occurs in living cells.



How many atoms are represented in the reactants of this equation?

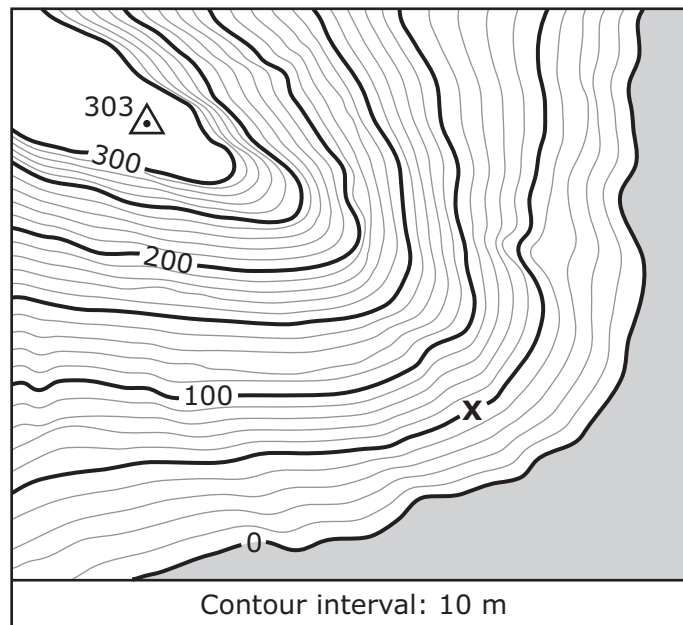
- F** 6
- G** 12
- H** 24
- J** 36

37 Which of the following best describes the velocity of an object?

- A 30 m/s
- B 30 m east
- C 30 m/s east
- D 30 m/s^2

38 A student stands at the point marked X on the topographic map below.

Topographic Map



What height in meters must the student climb in order to reach the top of the hill?

Record your answer and fill in the bubbles on your answer document. Be sure to use the correct place value.

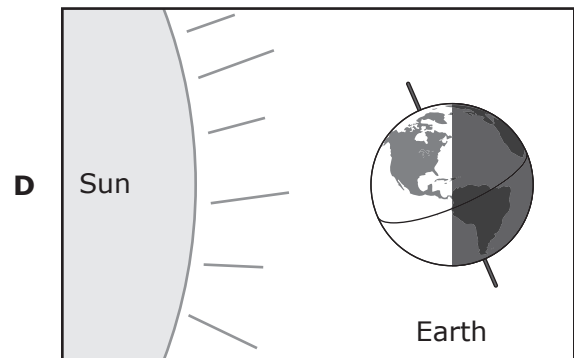
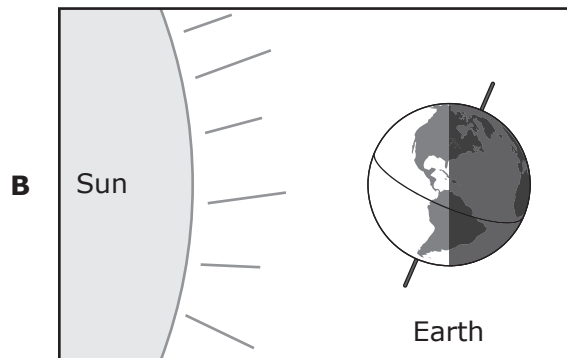
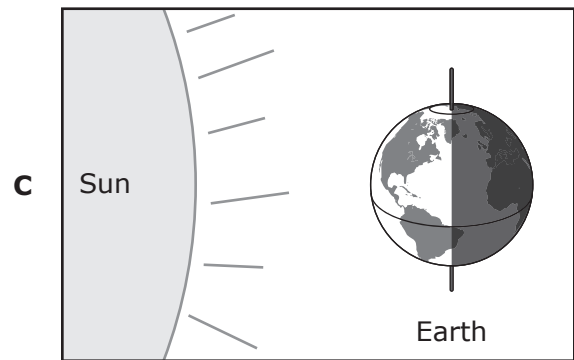
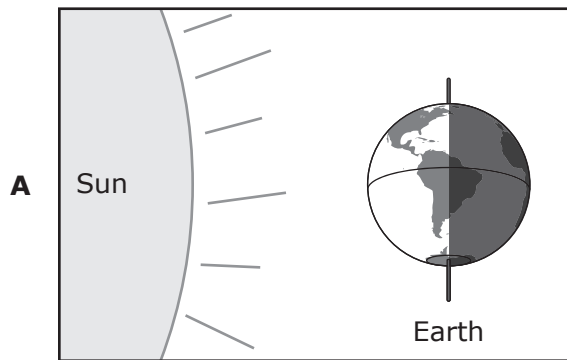
- 39** A chemist is identifying the elements present in a sample of seawater. What characteristic of an element's atoms always determines the element's identity?
- A** The number of protons
 - B** The number of neutrons
 - C** The location of valence electrons
 - D** The number of valence electrons

-
- 40** Copper compounds are used in large amounts to control the growth of algae and other aquatic plants. Copper interferes with photosynthesis in these organisms. What would be the most likely result of an accidental spill of these compounds in an aquatic environment?
- F** The fish would grow larger than usual.
 - G** The zooplankton population would increase.
 - H** The plants would increase in size but decrease in number.
 - J** The plant populations would be reduced.

- 41 In the summer many people enjoy going to the beach. The photograph shows a beach in Argentina, which is located in the Southern Hemisphere.

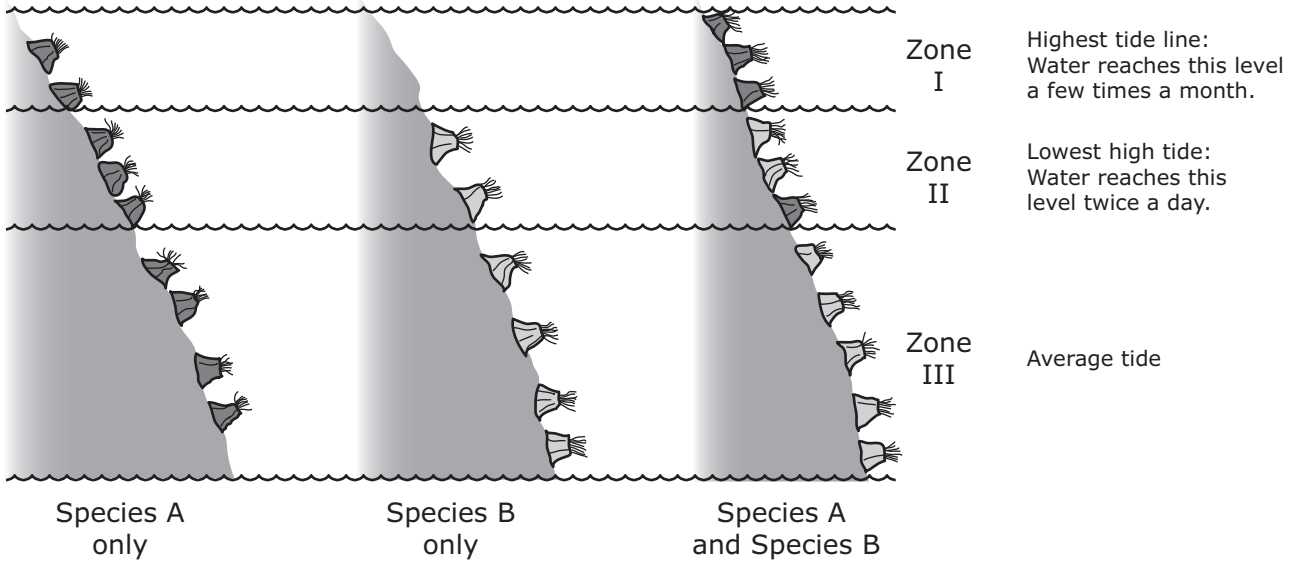


Which diagram shows the position of Earth when it is summer in Argentina?



- 42** Barnacles are marine organisms that live attached to surfaces as adults. A scientist studies two barnacle species that live on rocks on a coast in Scotland. In some rocky areas Species A is found alone. In other rocky areas Species B is found alone. In still other rocky areas both species are found together. The diagram below shows the two barnacle species on rocks in their habitats.

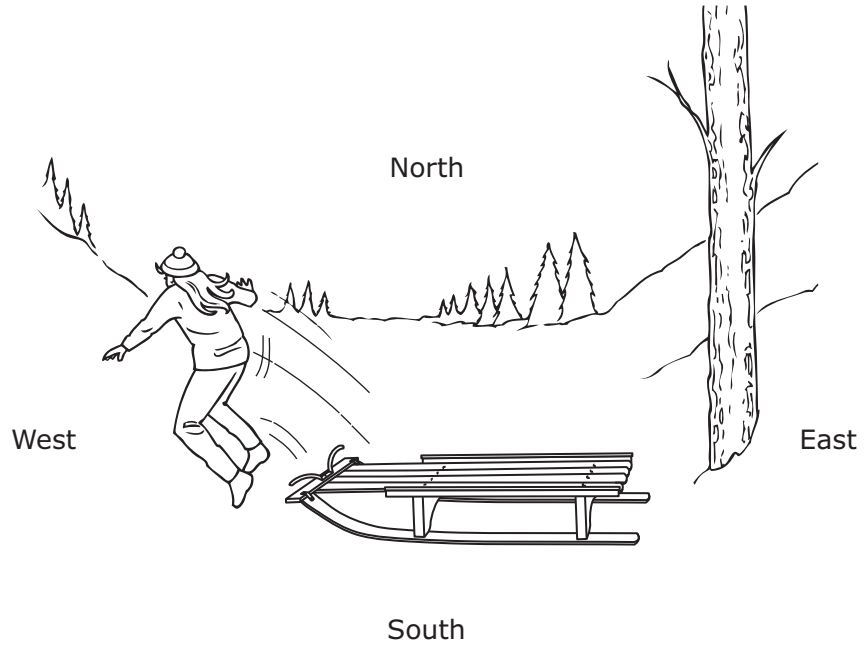
Two Species of Barnacles on a Coast in Scotland



If Species B were removed from the rocky areas it shares with Species A, Species A would most likely inhabit —

- F** Zone I only
- G** Zones I and II only
- H** Zone III only
- J** Zones I, II, and III

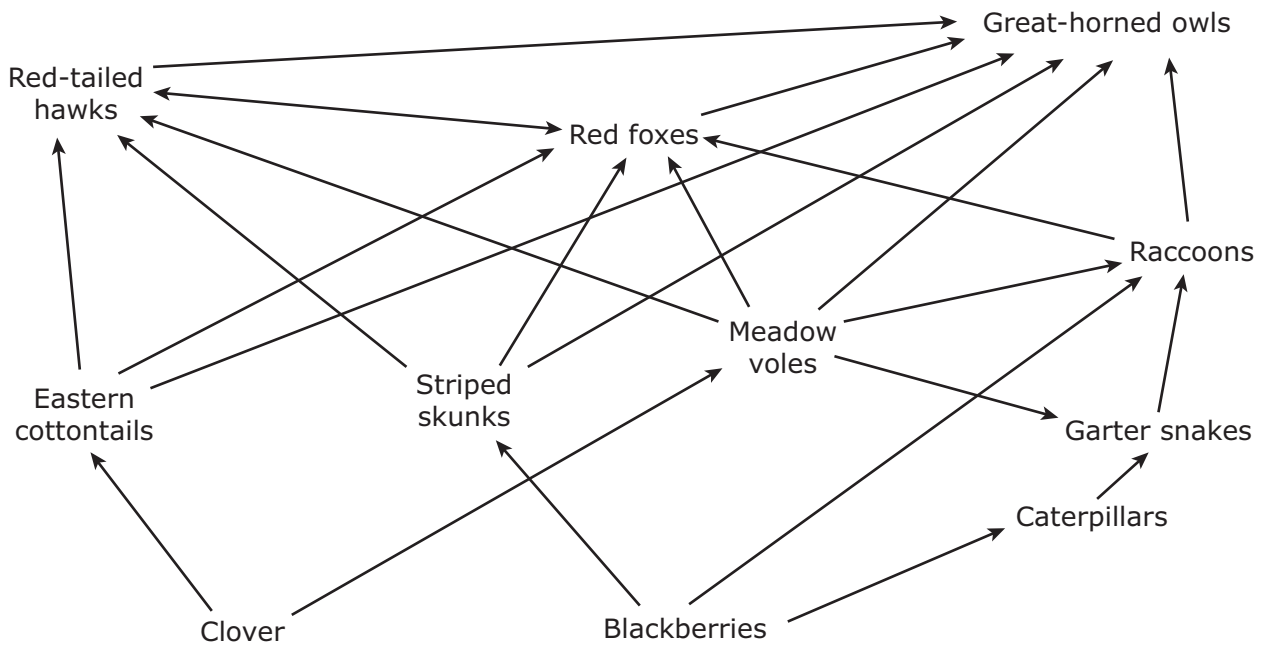
43 A student jumps off a sled toward the west after it stops at the bottom of an icy hill.



Based on the law of action–reaction, in what direction will the sled most likely move as the student jumps off?

- A North
- B South
- C East
- D West

44 Some students are studying the relationships between organisms in a food web.



How many predators shown in this food web are omnivores?

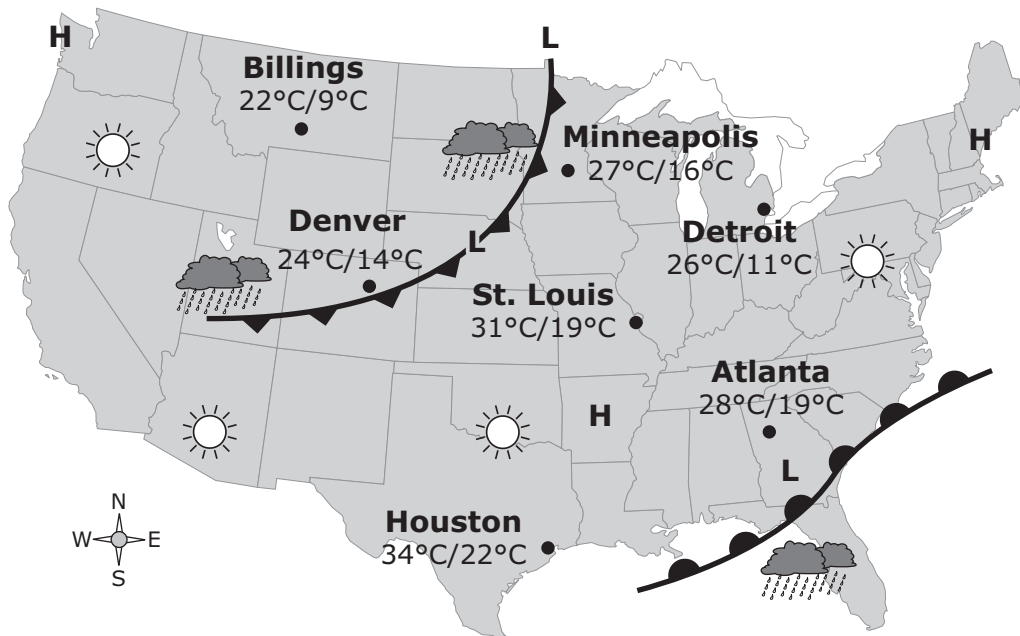
Record your answer and fill in the bubbles on your answer document. Be sure to use the correct place value.

45 In the early 1900s many scientists thought that an atom consisted of a positive substance with negative charges scattered throughout the substance. Then Ernest Rutherford completed an experiment that changed the concept of an atom. His discovery led to the understanding that an atom consists mostly of empty space with —

- A protons orbiting a dense nucleus made of electrons and neutrons
- B electrons orbiting a dense nucleus made of protons and neutrons
- C neutrons and protons orbiting a cloud of electrons
- D electrons and protons orbiting a cloud of neutrons

- 46** Tropical rain forests have the greatest biodiversity of any type of land ecosystem. How does biodiversity contribute to the sustainability of an ecosystem?
- F** The presence of more species with different adaptations makes it more likely that some organisms will survive an ecological disaster.
 - G** Greater genetic variation within species makes it more likely that some individuals will survive a disease outbreak.
 - H** The presence of a variety of herbivore species that can feed on a large number of different producer species helps ensure abundant prey for predators in the ecosystem.
 - J** All of the above

47 The weather map shows the weather conditions for one day across the United States.

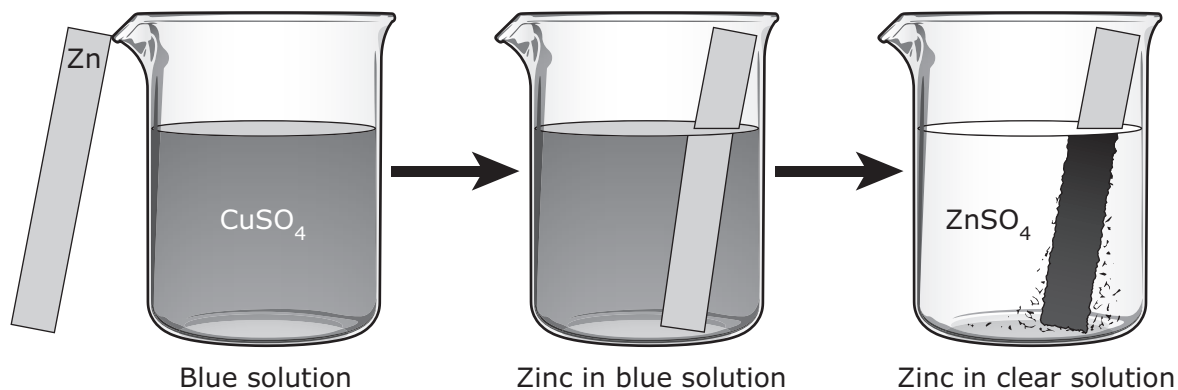


Based on the weather map, which city will most likely experience decreasing temperatures during the next 24 hours?

- A** Denver, because it is raining there
- B** Minneapolis, because a cold front is approaching
- C** Atlanta, because a warm front is approaching
- D** Houston, because it is in a high-pressure area

- 48 For an investigation a student poured a blue solution of CuSO_4 into a beaker. The student placed a shiny, silver-colored strip of zinc metal in the solution and observed the changes.

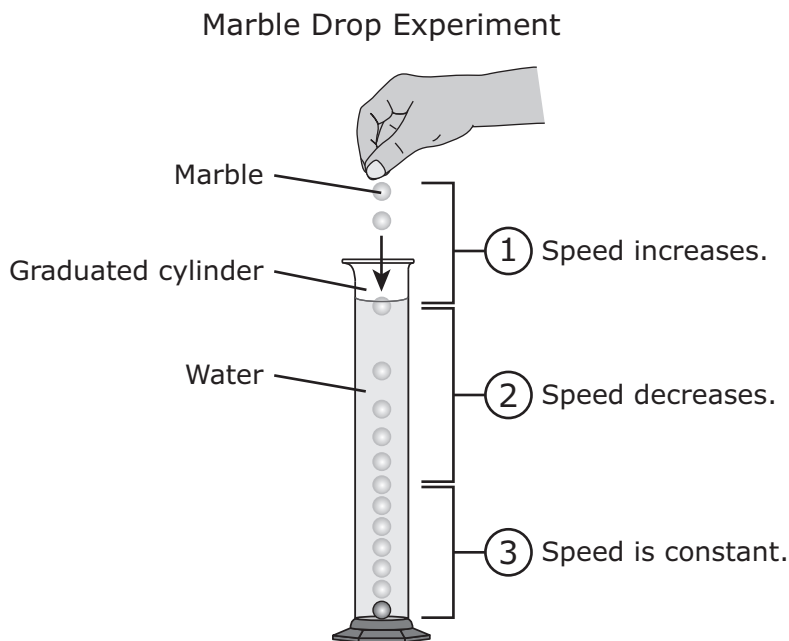
Observations of an Investigation



The student inferred that a chemical reaction occurred. What evidence supports this inference?

- F A dark solid formed on the zinc metal.
- G The zinc metal remained silver-colored and shiny.
- H The CuSO_4 solution turned blue when the zinc metal was added.
- J None of these

- 49** A student used a video camera to record another student dropping a marble through water in a graduated cylinder. The students watched the video in slow motion and made the observations shown below.



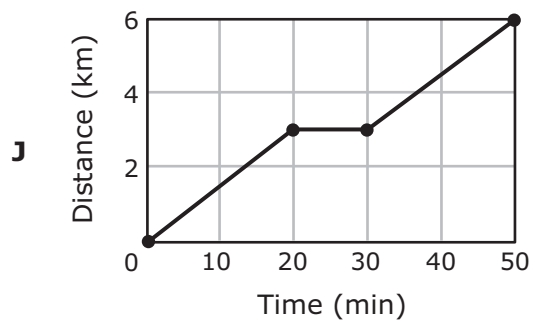
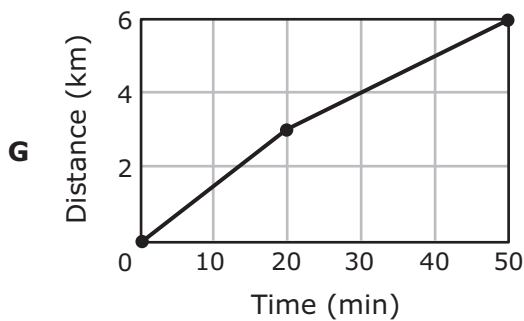
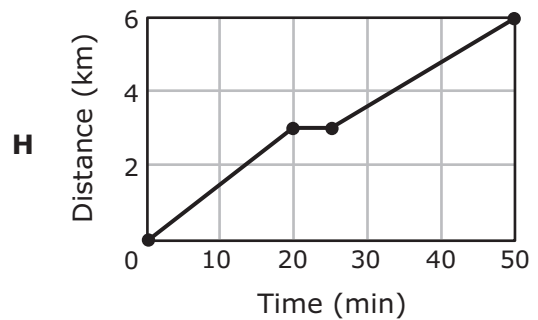
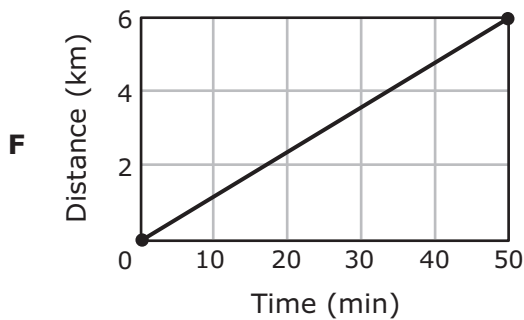
During which part or parts of the marble's fall did the marble experience unbalanced forces?

- A** Part 1 only
 - B** Parts 1 and 2 only
 - C** Part 3 only
 - D** Parts 2 and 3 only
-
- 50** Which of these correctly compares the masses of different objects in the universe?
- F** A moon has less mass than a star and more mass than the planet it orbits.
 - G** A planet has less mass than a galaxy and more mass than the star it orbits.
 - H** A galaxy has less mass than a moon and more mass than a planet.
 - J** A star has less mass than a galaxy and more mass than a planet.

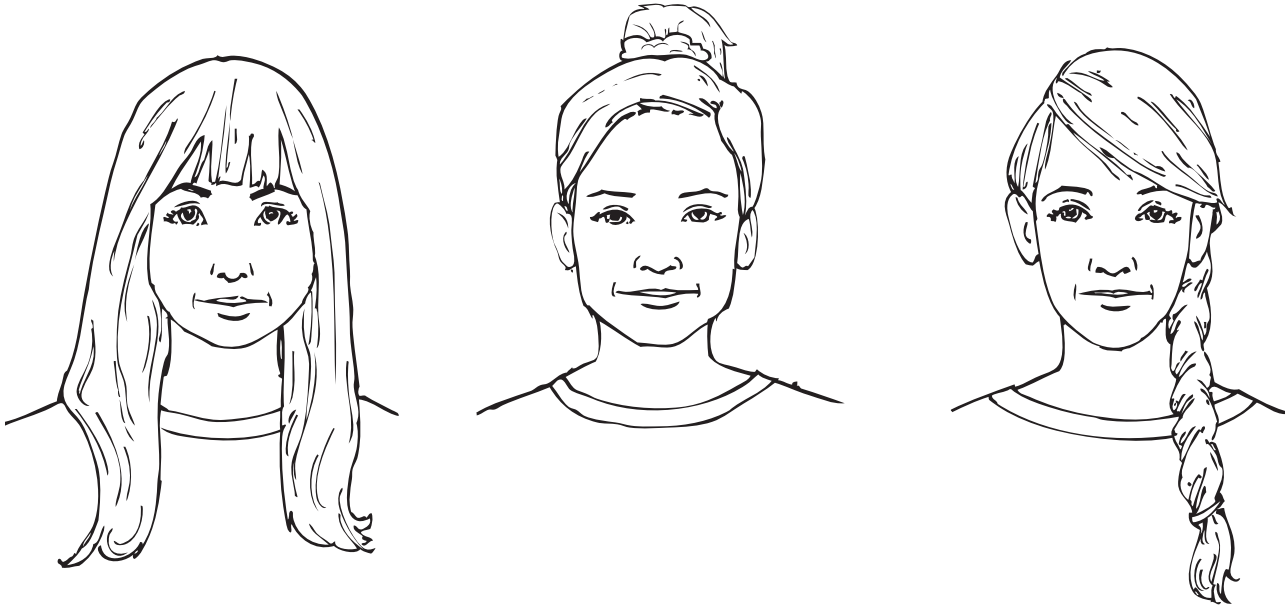
51 Which two elements on the periodic table are in the same period?

- A Sn and Rb
- B F and Cl
- C K and Ba
- D Se and Te

52 A student runs two times around a running path at a local park. Each lap is 3 km. The student completes the first lap in 20 minutes. The student then sits on a bench and rests for 5 minutes before completing the second lap in 25 minutes. Which graph best represents the student's motion?



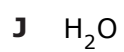
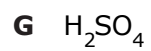
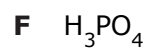
53 Three cousins have a similar appearance but different face shapes.



Which of these cell components are most involved in determining the basic shape of each girl's face?

- A** Genes, chromosomes, and nucleus
- B** Cytoplasm, chloroplasts, and genes
- C** Vacuoles, nucleus, and chromosomes
- D** Chromosomes, chloroplasts, and vacuoles

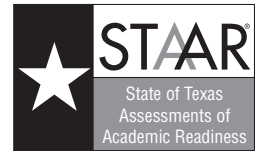
54 In which of these compounds are there twice as many oxygen atoms as hydrogen atoms?



BE SURE YOU HAVE RECORDED ALL OF YOUR ANSWERS
ON THE ANSWER DOCUMENT.



STAAR GRADE 8 SCIENCE REFERENCE MATERIALS



FORMULAS

$$\text{Density} = \frac{\text{mass}}{\text{volume}}$$

$$D = \frac{m}{V}$$

$$\text{Average speed} = \frac{\text{total distance}}{\text{total time}}$$

$$s = \frac{d}{t}$$

$$\text{Net force} = (\text{mass})(\text{acceleration})$$

$$F = ma$$

$$\text{Work} = (\text{force})(\text{distance})$$

$$W = Fd$$

STAAR GRADE 8 SCIENCE REFERENCE MATERIALS

PERIODIC TABLE OF THE ELEMENTS

																		18 8A
																		2 He 4.003 Helium
1 1A																		17 7A
1 H 1.008 Hydrogen																		16 6A
2 2A																		15 5A
3																		14 4A
4																		13 3A
5																		12 2B
6																		11 1B
7																		10
8																		9
9																		8
10																		7
11																		6
12																		5
13																		4
14																		3
15																		2
16																		1
17																		0
18																		0

Atomic number	14
Symbol	Si
Atomic mass	28.086
Name	Silicon

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1A	2A	3B	4B	5B	6B	7B	8B	9	10	11B	12B	13A	14A	15A	16A	17A	18
H 1.008 Hydrogen	He 4.003 Helium	Li 6.941 Lithium	Be 9.012 Beryllium	B 10.812 Boron	C 12.011 Carbon	N 14.007 Nitrogen	O 15.999 Oxygen	F 18.998 Fluorine	Ne 20.180 Neon	Na 22.990 Sodium	Mg 24.305 Magnesium	Al 26.982 Aluminum	Si 28.086 Silicon	P 30.974 Phosphorus	S 32.066 Sulfur	Cl 35.453 Chlorine	Ar 39.948 Argon
K 39.098 Potassium	Ca 40.078 Calcium	Sc 44.956 Scandium	Ti 47.867 Titanium	V 50.942 Vanadium	Cr 51.996 Chromium	Mn 54.938 Manganese	Fe 55.845 Iron	Co 58.933 Cobalt	Ni 58.693 Nickel	Cu 63.546 Copper	Zn 65.38 Zinc	Ga 69.723 Gallium	Ge 72.64 Germanium	As 74.922 Arsenic	Se 78.96 Selenium	Br 79.904 Bromine	Kr 83.798 Krypton
Rb 85.468 Rubidium	Sr 87.62 Strontium	Y 88.906 Yttrium	Zr 91.224 Zirconium	Nb 92.906 Niobium	Mo 95.96 Molybdenum	Tc (98) Technetium	Ru 101.07 Ruthenium	Rh 102.906 Rhodium	Pd 106.42 Palladium	Ag 107.868 Silver	Cd 112.412 Cadmium	In 114.818 Indium	Sn 118.711 Tin	Sb 121.760 Antimony	Te 127.60 Tellurium	I 126.904 Iodine	Xe 131.294 Xenon
Cs 132.905 Cesium	Ba 137.328 Barium	Lu 174.967 Lutetium	Hf 178.49 Hafnium	Ta 180.948 Tantalum	W 183.84 Tungsten	Re 186.207 Rhenium	Os 190.23 Osmium	Ir 192.217 Iridium	Pt 195.085 Platinum	Au 196.967 Gold	Hg 200.59 Mercury	Tl 204.383 Thallium	Pb 207.2 Lead	Bi 208.980 Bismuth	Po (209) Polonium	At (210) Astatine	Rn (222) Radon
Fr (223) Francium	Ra (226) Radium	Lr (262) Lawrencium	Rf (267) Rutherfordium	Db (268) Dubnium	Sg (271) Seaborgium	Bh (272) Bohrium	Hs (270) Hassium	Mt (276) Meitnerium	Ds (281) Darmstadtium	Rg (280) Roentgenium							

Mass numbers in parentheses are those of the most stable or most common isotope.

Lanthanide Series										Actinide Series																	
57	La 138.905 Lanthanum	58	Ce 140.116 Cerium	59	Pr 140.908 Praseodymium	60	Nd 144.242 Neodymium	61	Pm (145) Promethium	62	Sm 150.36 Samarium	63	Eu 151.964 Europium	64	Gd 157.25 Gadolinium	65	Tb 158.925 Terbium	66	Dy 162.500 Dysprosium	67	Ho 164.930 Holmium	68	Er 167.259 Erbium	69	Tm 168.934 Thulium	70	Yb 173.055 Ytterbium
89	Ac (227) Actinium	90	Th 232.038 Thorium	91	Pa 231.036 Protactinium	92	U 238.029 Uranium	93	Np (237) Neptunium	94	Pu (244) Plutonium	95	Am (243) Americium	96	Cm (247) Curium	97	Bk (247) Berkelium	98	Cf (251) Californium	99	Es (252) Einsteinium	100	Fm (257) Fermium	101	Md (258) Mendelevium	102	No (259) Nobelium

**STAAR
GRADE 8
Science
April 2015**